

# Photocatalytic reduction of CO<sub>2</sub> with H<sub>2</sub>O into C<sub>2</sub>H<sub>6</sub> mediated by dual metalation strategy

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Artificial solar-energy-driven CO<sub>2</sub> reduction into value-added fuels under mild conditions provides a promising strategy to mitigate energy crisis and alleviate environmental issues around the world. Among diverse products of CO<sub>2</sub> reduction reaction, C<sub>2+</sub> compounds are more desirable in terms of their high energy density and commercial value [1]. Thus far, various photocatalysts have been constructed for CO<sub>2</sub> reduction. Nevertheless, the conventional photocatalytic systems, even the most advanced ones, primarily produce C<sub>1</sub> products, bearing with the shortcomings of low activity, low efficiency, and also, demanding sacrificial agents [2]. Thereby, the development of practical photocatalytic devices for converting CO<sub>2</sub> into C<sub>2+</sub> products without any applied bias or sacrificial agents is of enormous scientific significance [3,4]. However, the exploration of efficient photocatalysts incorporated with suitable semiconductors to facilitate C–C coupling remains a huge challenge [5].

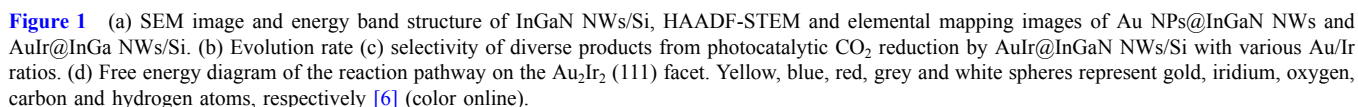
Very recently, reporting in *Nature Catalysis*, Mi, Song, Zhou and co-workers fabricated a state-of-the-art artificial photosynthesis-integrated device (APID), namely, the AuIr dual-metalated composite accurately photodeposited on the III-nitride semiconductor of InGa<sub>0.56</sub>N NWs/Si, noted as, AuIr@InGa<sub>0.56</sub>N NWs/Si [6]. By employing CO<sub>2</sub>, water and light as the only inputs, it can realize highly efficient photocatalytic CO<sub>2</sub> reduction to C<sub>2</sub>H<sub>6</sub> in water with no bias or sacrificial agents (Figure 1a). Specifically, under con-

centrated light irradiation of 3.5 W cm<sup>−2</sup>, the designed device of Au<sub>0.44</sub>Ir<sub>0.56</sub>@InGa<sub>0.56</sub>N NWs/Si can reduce CO<sub>2</sub> towards C<sub>2</sub>H<sub>6</sub> with the activity up to 58.8 mmol g<sup>−1</sup> h<sup>−1</sup> (Figure 1b), achieving a turnover number of 54,595 and a selectivity of 5.6% (Figure 1c), exhibiting no significant decrease for the performance within 60 h. Meanwhile, the generation rate of CH<sub>4</sub> is 125.4 mmol g<sup>−1</sup> h<sup>−1</sup>, which is several orders of magnitude higher than that of the reported catalytic systems to date, leading to a total selectivity of 17.6% for hydrocarbons (Figure 1c). Moreover, considering the useful by-products of syngas (CO and H<sub>2</sub>), the designed device possesses a great light-to-fuel (LTF) efficiency of approximate 0.59%. These above results demonstrate a tremendous superiority for the optimized AuIr dual metalation strategy in the sphere of photocatalytic CO<sub>2</sub> reduction [7]. Of note, H<sub>2</sub>O<sub>2</sub> was simultaneously formed from water oxidation by photoexcited holes, making this process more commercially promising.

The authors found that due to the epitaxial growth of 1D InGa<sub>0.56</sub>N NWs on 3-inch silicon wafers, the pre-synthesized semiconductor displayed outstanding optical absorption and charge transfer efficiency for promoting charge separation, which offered an ideal platform to support the bimetallic AuIr alloy for promoting CO<sub>2</sub> reduction towards C<sub>2+</sub> product. Additionally, the authors validated that it was iridium but not gold in the AuIr alloy responsible for the C–C bond generation. Meanwhile, the existence of electronic interactions between gold and iridium was also verified, which can extremely enhance C–C coupling during the catalytic process.

According to the control experiments, it is found that while

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results preliminary suggest that the CO<sub>2</sub> insertion into –CH<sub>3</sub> should be the determining step for C–C coupling. Furthermore, the results of density functional theory (DFT) calculations on the Au<sub>2</sub>Ir<sub>2</sub> (111) facet demonstrate that the reaction pathway would go through the hydrogenation of CO<sub>2</sub> to \*COH and then to \*C→\*CH→\*CH<sub>2</sub>→\*CH<sub>3</sub>, subsequently,

the insertion of  $^*\text{CO}_2$  into  $^*\text{CH}_3$ , forming the main  $^*\text{CH}_3\text{COO}$  intermediate (Figure 1d). Finally, established on the above investigations, the authors proposed that C–C coupling can be efficiently accomplished by inserting  $\text{CO}_2$  into  $^*\text{CH}_3$  due to Ir modulating Au with the best Au:Ir ratio of around 1:1 in AuIr@InGaN NWs/Si.

In this work, taking advantage of the synergistic effect of gold and iridium, as well as the one-dimensional InGaN nanowire semiconductor with unique optical and electronic properties, the well-designed photocatalytic device exhibits a break-through performance for generating  $\text{C}_2\text{H}_6$  derived from  $\text{CO}_2$  reduction, which reveals an inspiring strategy for designing novel robust photocatalysts for C–C coupling *via* benign and environmentally-friendly approach. The finding of this study opens a door for the production of  $\text{C}_{2+}$  compounds from sunlight,  $\text{CO}_2$  and water without any other energy input, which paves the way for realizing  $\text{CO}_2$  resource utilization through photocatalysis. Of note, apart from  $\text{C}_{2+}$  fuels yielded from photocatalytic  $\text{CO}_2$  reduction, the concurrent formation of  $\text{H}_2\text{O}_2$  from photocatalytic water

oxidation over AuIr@InGaN NWs/Si makes such an artificial photosynthesis process more commercially promising.

**Conflict of interest** The authors declare no conflict of interest.

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